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**Fourth Semester B.E. Degree Examination, Dec.2013/Jan.2014**  
**Applied Thermodynamics**

Time: 3 hrs.

Max. Marks:100

- Note:1. Answer FIVE full questions, selecting at least TWO questions from each part.**  
**2. Use of thermodynamics data handbook permitted.**

PART – A

- 1 a. Explain the following: i) Air fuel ratio ii) Enthalpy of formation. (06 Marks)  
 iii) Combustion efficiency. (08 Marks)
- b. With a neat sketch, explain the analysis of exhaust gases by Orsat apparatus. (08 Marks)
- c. A producer gas used as a fuel has the following volumetric composition:  
 $H_2 = 28\%$ ,  $CO = 12\%$ ,  $CH_4 = 2\%$ ,  $CO_2 = 16\%$ ,  $N_2 = 42\%$   
 Find the volume of air required for complete combustion of  $1\text{ m}^3$  of this gas. Air contains 21% by volume of oxygen. (06 Marks)
- 2 a. With a help of P-V and T-S diagram of the air standard dual combustion cycle. Derive the expression for the ideal efficiency. (10 Marks)
- b. The stroke and cylinder diameter of a compression ignition engine are 250 mm and 150 mm respectively. If the clearance volume is  $0.0004\text{ m}^3$  and fuel injection takes place at constant pressure for 5 percent of the stroke. Determine the efficiency of the engine. Note the engine is working on the diesel cycle. (10 Marks)
- 3 a. Explain any one method employed for improvement of thermal efficiency of open cycle gas turbine plant. (08 Marks)
- b. State the fundamental difference between the jet propulsion and rocket propulsion. (04 Marks)
- c. In a gas turbine the compressor takes in air at a temperature of  $15^\circ\text{C}$  and compresses it to four times the initial pressure with an isentropic efficiency of 82%. The air is then passed through a heat exchanger heated by the turbine exhaust before reaching the combustion chamber. In the heat exchanger 78% of the available heat is given to air. The maximum temperature after constant pressure combustion is  $600^\circ\text{C}$  and the efficiency of the turbine is 70%. The working fluid have the characteristics of air. Neglect all losses except those mentioned. Find the efficiency of the cycle. Take  $R = 0.287 \frac{\text{KJ}}{\text{kgK}}$  and  $\gamma = 1.4$ . (08 Marks)
- 4 a. Derive an expression for modified Rankine cycle with P-V and T-S diagram. (08 Marks)
- b. Comment Carnot cycle cannot be used practically. (02 Marks)
- c. A turbine is supplied with steam at a pressure of 32 bar and a temperature of  $410^\circ\text{C}$ . The steam then expands isentropically to a pressure of 0.08 bar. Find the dryness fraction of the end of expansion and thermal efficiency of the cycle. If the steam is reheated at 5.5 bar to a temperature of  $395^\circ\text{C}$  and then expands isentropically to a pressure of 0.08 bar. What will be the dryness fraction and thermal efficiency of the cycle? (10 Marks)

PART – B

- 5 a. In a two stage air compressor, prove that the work done is minimum with perfect intercooling when  $P_2 = \sqrt{P_1 P_3}$ . (06 Marks)

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- 5 b. Define the following:
- Volumetric efficiency.
  - Isentropic efficiency.
  - Mechanical efficiency. (06 Marks)
- c. A two stage air compressor operates at 200 rpm and delivers  $2 \text{ m}^3/\text{min}$  at suction condition. At suction the temperature and pressure are  $27^\circ\text{C}$  and  $0.1 \text{ MPa}$  respectively. The air is delivered at  $5 \text{ MPa}$ , the clearance of both L.P and H.P cylinder is 5%. Assuming perfect intercooling and a constant compression and expansion index of 1.25. Determine:
- Minimum power required.
  - Volumetric efficiency of both L.P. and H.P. cylinders. (08 Marks)
- Take  $R = 0.287 \text{ KJ/kgK}$ .

- 6 a. Define the following: i) COP. ii) Ton of refrigeration. (06 Marks)
- b. Explain the working principle of a vapour absorption refrigeration system. (06 Marks)
- c. An ammonia ice plant operates between a condenser temperature of  $30^\circ\text{C}$  and an evaporator temperature of  $-5^\circ\text{C}$ . It produces 10 tons of ice per day from water at  $30^\circ\text{C}$  to ice at  $-5^\circ\text{C}$ . Assume the state at the end of compression process as dry saturated. Determine,
- Mass flow rate of refrigerant.
  - Capacity of refrigerant plant.
  - COP.
- Take :  $C_p$  of water =  $4186 \text{ J/kgK}$   
 Specific heat of ice =  $194 \text{ J/kgK}$   
 L.T of fusion at  $0^\circ\text{C}$  for water =  $335 \times 10^3 \text{ J/kg}$

Saturation data for ammonia ( $\text{NH}_3$ )

Temp = $^\circ\text{C}$	Sat.Pr. KPa	SP = Volume $\text{m}^3/\text{kg}$		Enthalpy $\text{KJ/kgK}$		Entropy $\text{KJ/kgK}$	
		Vf	Vg	hf	hg	Sf	Sg
$30^\circ$	1169.49	0.00168	0.1106	322.9	1467.9	1.2028	4.9805
$-15^\circ$	236.48	0.002518	0.5091	112.35	1425.7	0.4544	5.545

- 7 a. Define the following with respect to air-water vapour mixture:
- Specific humidity
  - Wet bulb temperature
  - Relative humidity
  - Adiabatic saturation temperature
  - Dew point temperature. (10 Marks)
- b. Air at  $30^\circ\text{C}$ , 60% RH,  $0.8808 \text{ m}^3/\text{kg}$  specific volume,  $71.2 \text{ KJ/kg}$  enthalpy, flows over a cooling coil at  $250 \text{ m}^3/\text{min}$ . From the chart the corresponding dew point is  $21.5^\circ\text{C}$ . After flowing over the coil the temperature reduces to  $23^\circ\text{C}$ ; with  $64.2 \text{ KJ/kg}$  enthalpy:
- Sketch the process on a psychrometric chart.
  - Find the refrigeration capacity of the coil in tons. (10 Marks)
- 8 a. Explain how Morse test will help you in finding the indicated power of the multi-cylinder engine. (08 Marks)
- b. The following observations were recorded in a test of one hour duration on a single cylinder oil engine working on four stroke cycle:
- Bore =  $300 \text{ mm}$ , Stroke =  $450 \text{ mm}$ , Average speed =  $200 \text{ rpm}$ ,  
 m.e.p =  $5.8 \text{ bar}$ , Calorific value of fuel =  $41800 \text{ KJ/kg}$ ,  
 Mass of fuel (used) =  $8.8 \text{ kg}$ , Brake friction load =  $1860 \text{ N}$   
 Quality of cooling water =  $650 \text{ kg}$ , Temperature rise =  $22^\circ\text{C}$   
 Diameter of brake wheel =  $1.22 \text{ m}$   
 Determine : i) Brake thermal efficiency  
 ii) Mechanical efficiency.  
 Draw the heat balance sheet. (12 Marks)